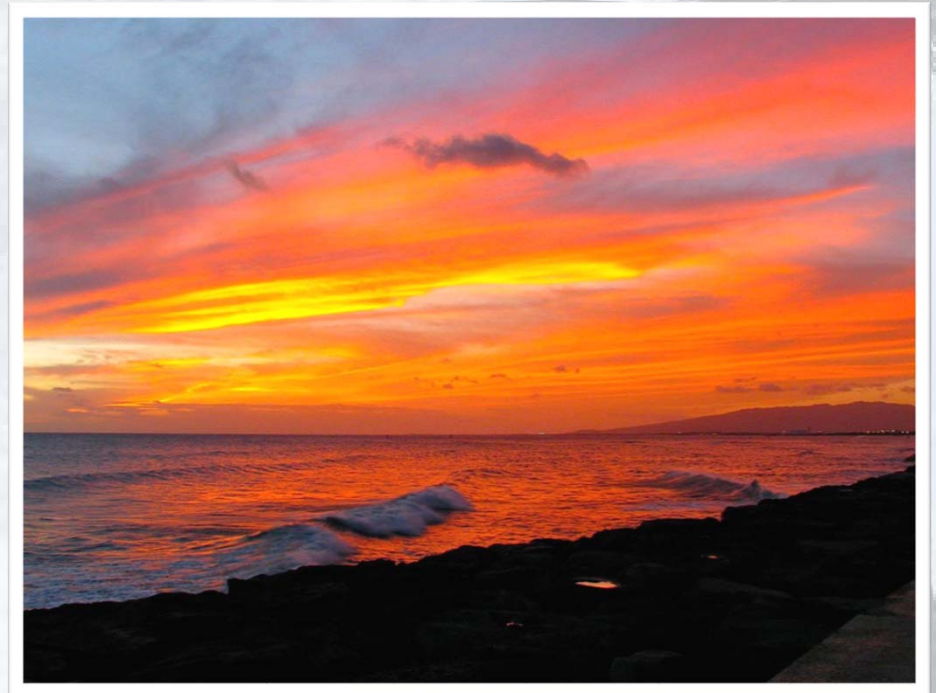
The background of the slide is a photograph of a tropical landscape. It features two palm trees, one on the left and one on the right, silhouetted against a bright blue sky filled with soft, white, scattered clouds. The overall atmosphere is bright and clear.

MET 102 Pacific Climates and Cultures

Lecture 5: Water and Rising Air

Discussion – Ackerman & Knox

- **What are the different ways of measure water vapor in the air?**
 - Mixing Ratio
 - Vapor Pressure
 - Relative Humidity
 - Dew Point/Frost Point
- **What are the 4 main ways are for make air rise?**
 - Orographic Lifting
 - Frontal Wedging
 - Convection (Differential Heating)
 - Convergence



Discussion – Ackerman & Knox

We're actually going to talk about these in depth on Thursday:

- **What are the main ways we classify clouds?**

- Shape
- Height in the atmosphere
- Whether or not they rain!



- **Do you think fog happens often on the Pacific Islands? Why or why not?**

- Fog can happen anywhere if the temperature and moisture content allow.
- The water and air may not get cold enough on most islands to allow for fog formation, however, valley fog on high islands is a possibility 😊

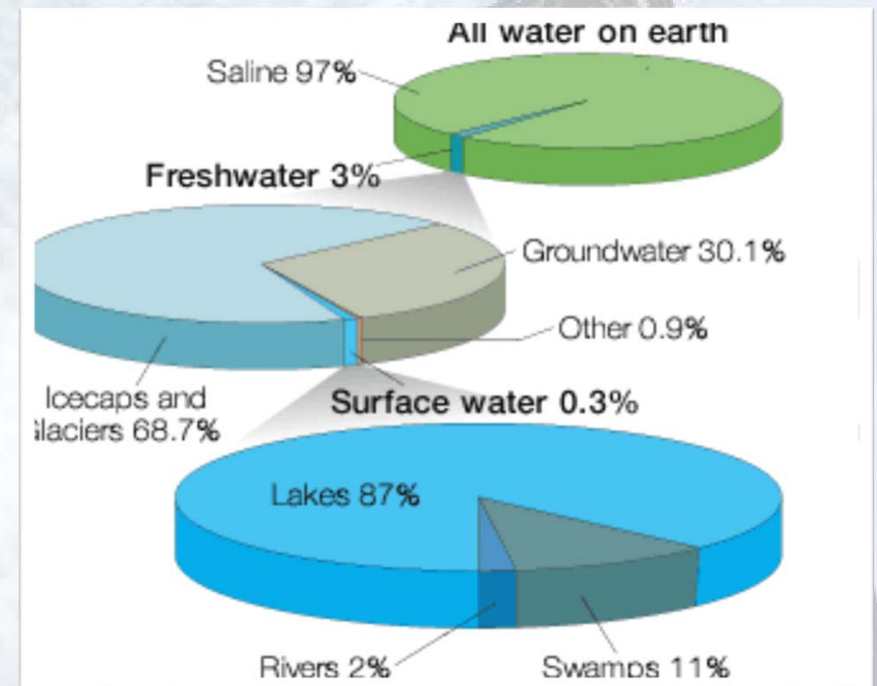
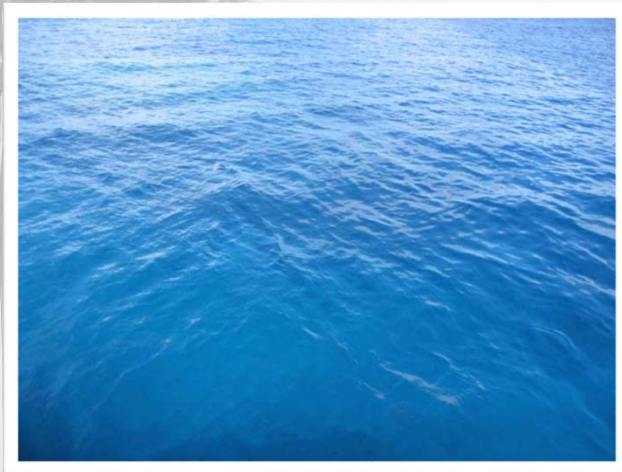
Water is Everywhere

- Oceans
- Rivers/Streams/Waterfalls
- Pools/Lakes
- Air
- Soil
- Living tissue
- Snow/Ice/Glaciers



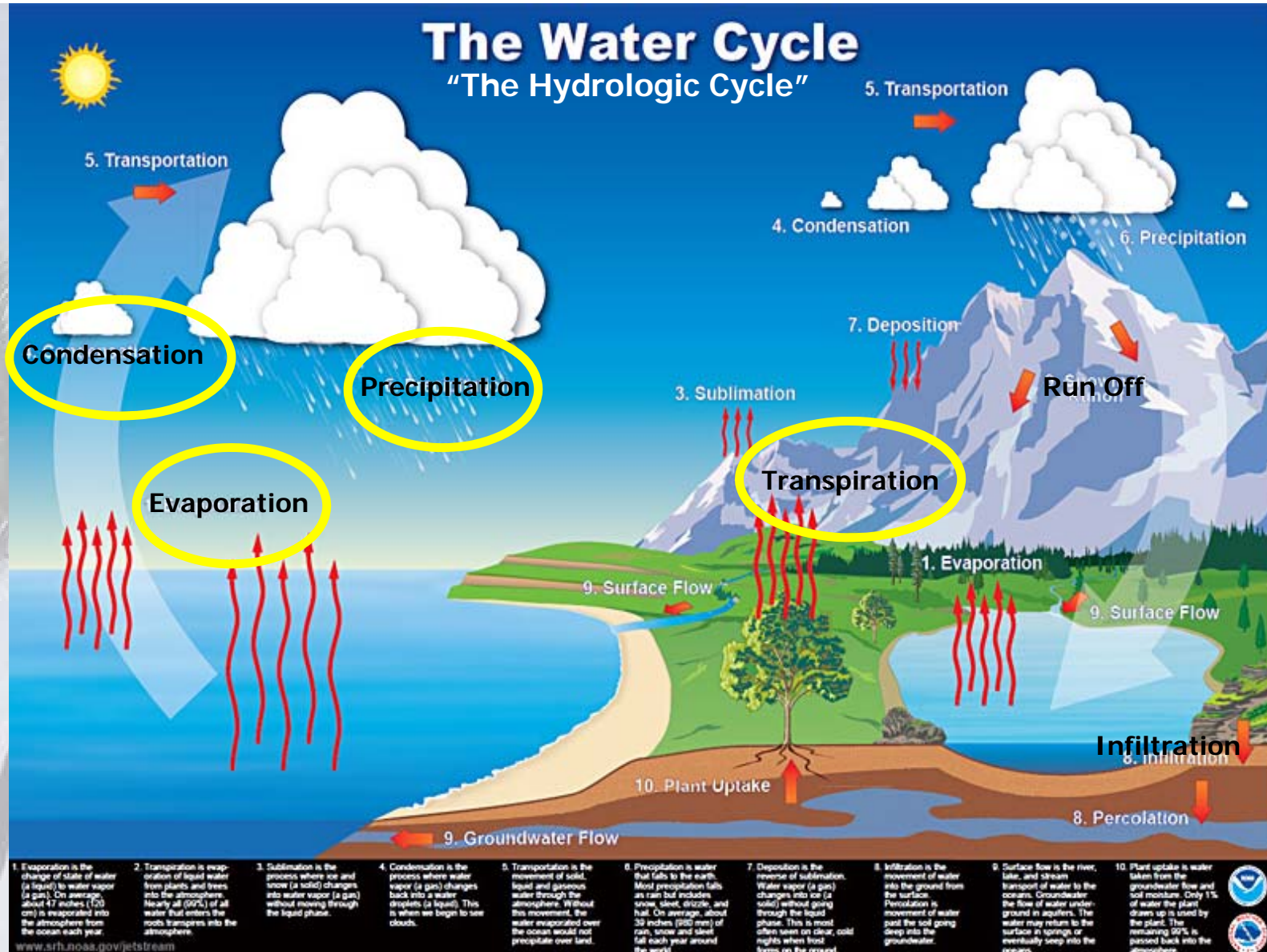
Water, water, everywhere, but not a drop to drink....

- Oceans account for most of water (>97%)
 - Not readily useable by humans or plants
- Ice sheets in Antarctica and Greenland (~3%)
- Atmosphere has only a little (0.001%)



The Water Cycle

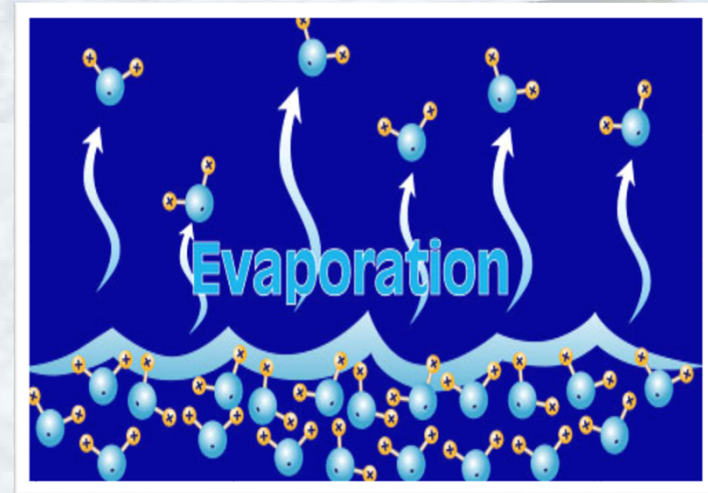
"The Hydrologic Cycle"



Processes Involving Water in the Atmosphere

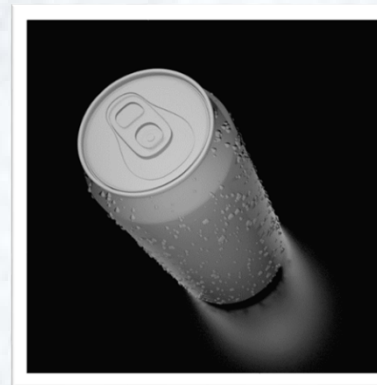
- **Evaporation – Requires Energy**

- The process by which a liquid is transformed into a gas
- Powered by the Sun!
- Solar Radiation heat up the water molecules until they are “freed” from the liquid state
- Heat is absorbed during evaporation



- **Condensation – Releases Energy**

- The change from a gas to a liquid
- Responsible for the formation of clouds
- Heat is released during condensation

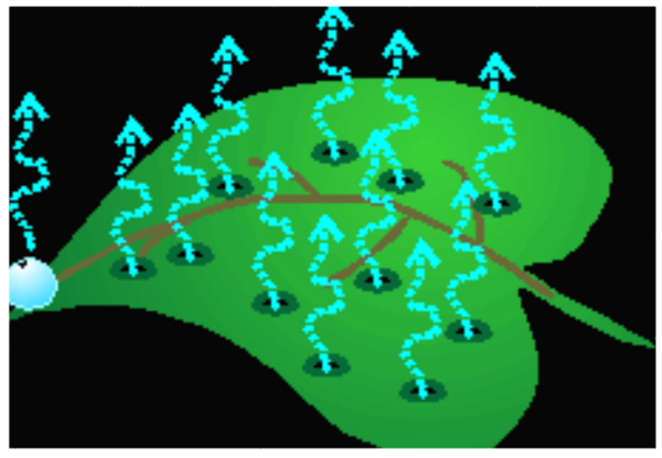


Processes Involving Water in the Atmosphere



• Precipitation – Over Land & Oceans

- Falling liquid or solid in the atmosphere.
- Returned the water to the ocean or soaks into the ground
- Balances Evaporation
 - Average annual precipitation equals evaporation.



• Transpiration

- The release of water vapor to the atmosphere by plants
- Plants uptake water through their roots that fell as precipitation
- Not as important as evaporation

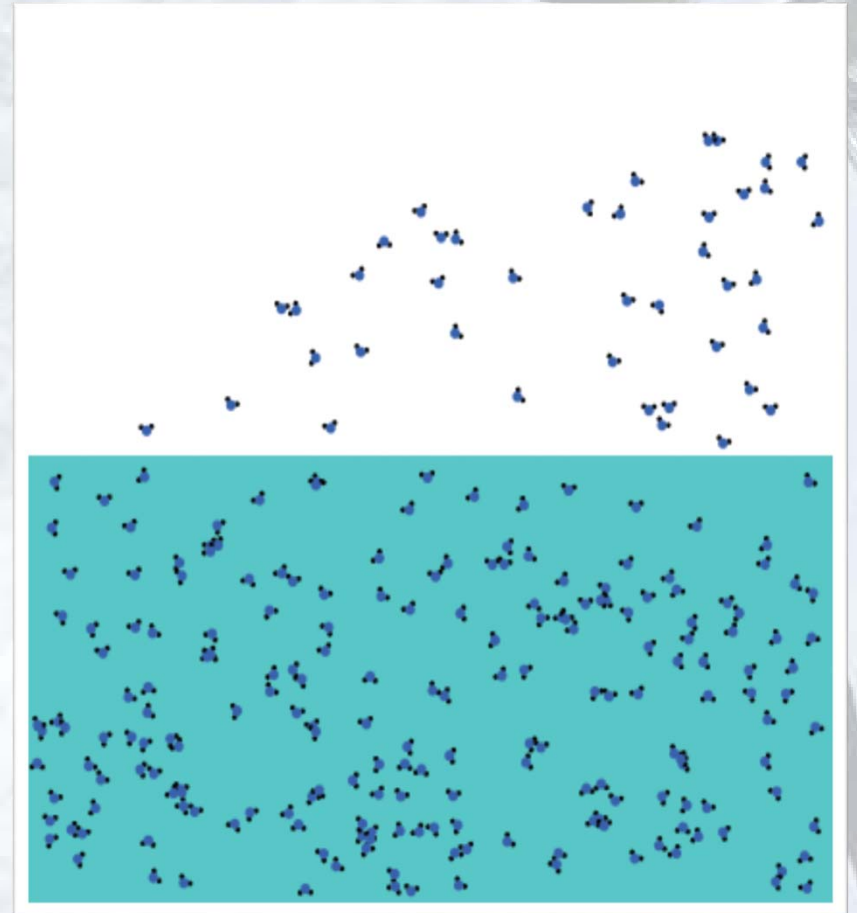
Water Vapor Content of the Air

- **Humidity**

- The general term used to describe the **amount** of **water vapor** in the air

- **Other Ways of thinking about it:**

- Mixing Ratio
- Vapor Pressure
- Relative Humidity
- Dew Point/Frost Point



Mixing Ratio

- The **MASS** of water vapor in a unit of air compared to the remaining **MASS of dry air**.
- **NOT** affected by changes in temperature and pressure!!!
- **MASS** doesn't change as a function of temperature.

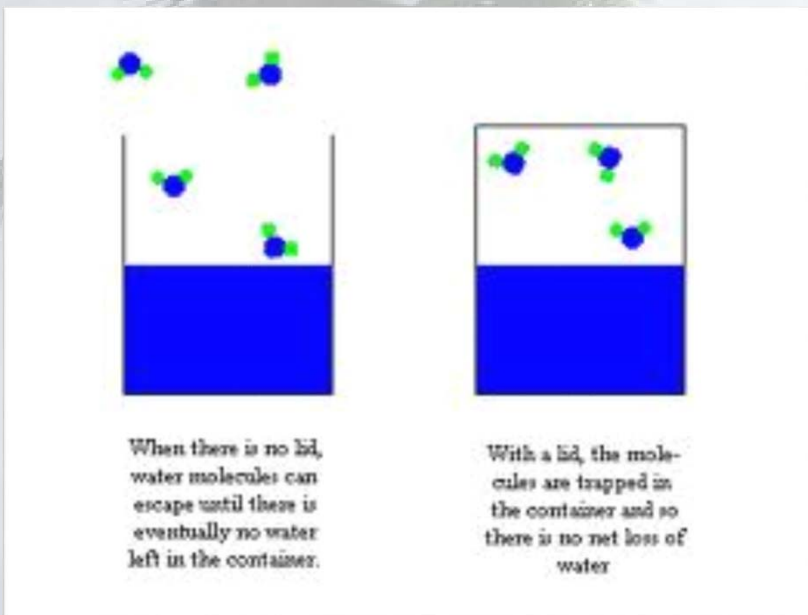
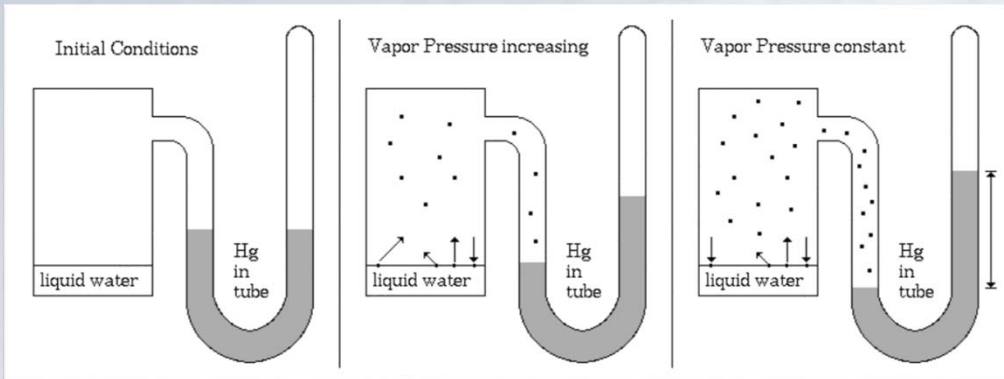
$$\text{Mixing Ratio} = \frac{\text{Mass of water vapor (grams)}}{\text{Mass of dry air (kilograms)}}$$

Vapor Pressure

- That part of the **total atmospheric pressure** attributable to its water-vapor content.
- As **more water vapor** is added to dry air the vapor **pressure increases**.
- **MASS** doesn't change as a function of temperature.

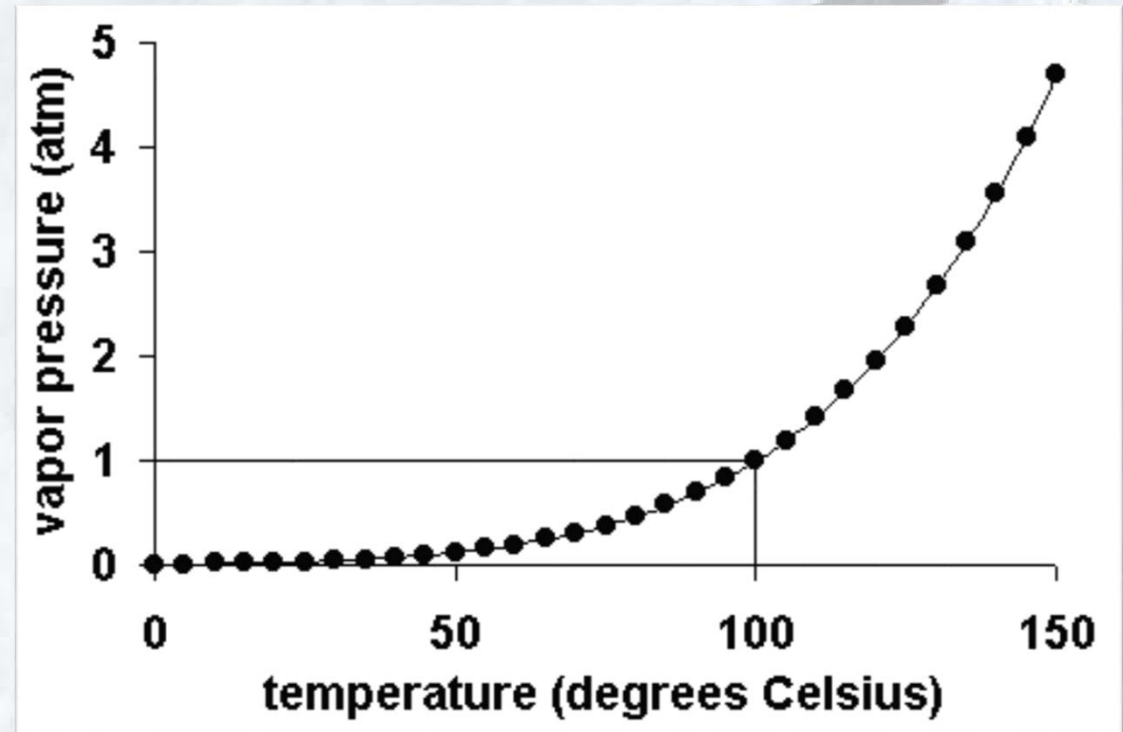
Initially more molecules leave the surface of the water than return. Over time: **number of molecules leaving = the number molecules returning**

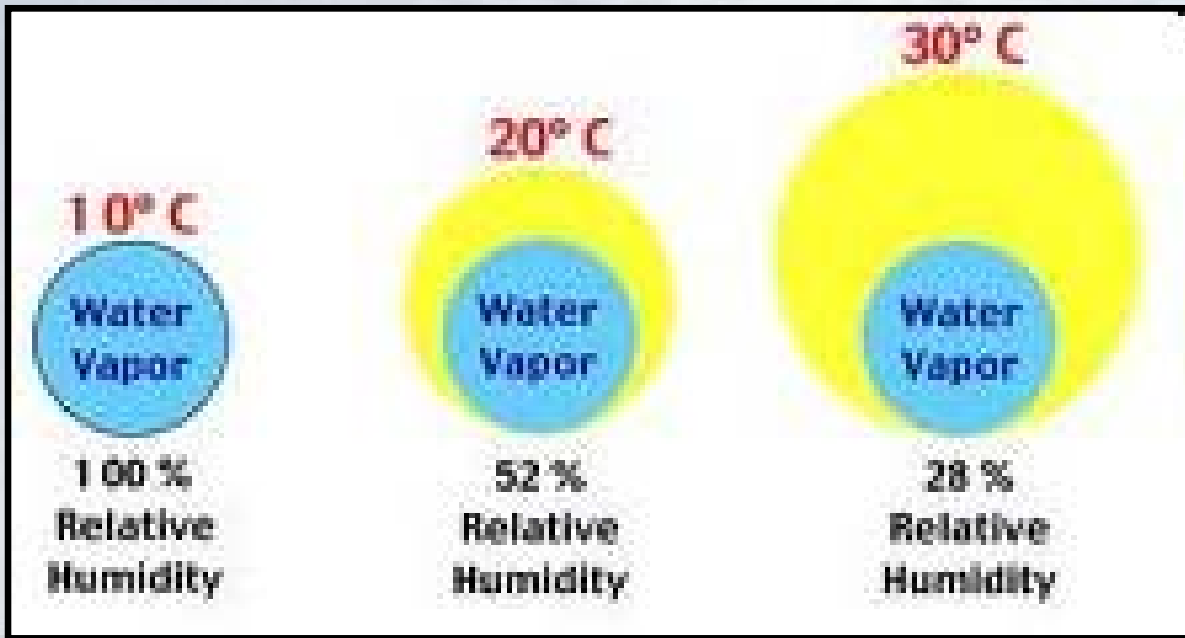
This is **SATURATION**



Saturation Vapor Pressure

- When air is saturated the pressure exerted by the motion of the water vapor molecules is called the **Saturation Vapor Pressure**.
- Varies as a function of **temperature**
- You can “**FIT**” more water vapor in warmer air.



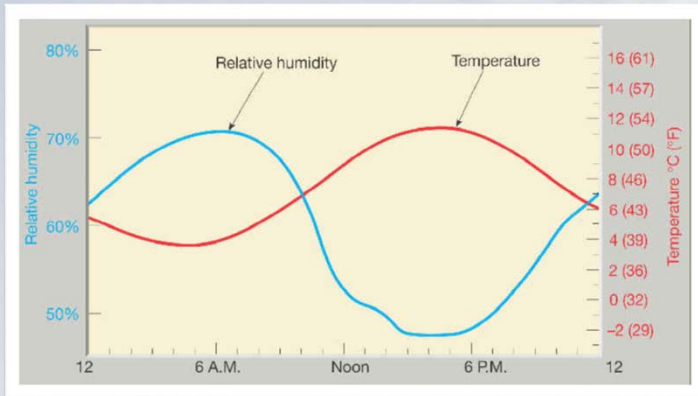


Relative Humidity

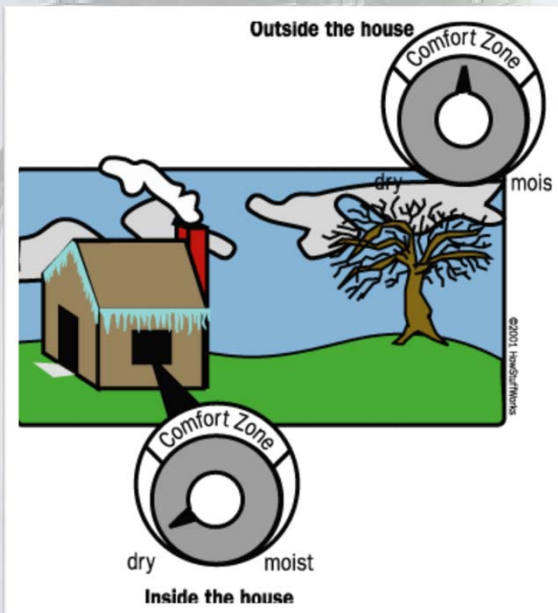
- The ratio of the air's **actual water vapor content** compared with the **amount of water vapor required for saturation** at that **temperature** and **pressure**

$$\text{Relative Humidity} = \frac{\text{Water vapor content}}{\text{Water vapor capacity}} \times 100 \text{ percent}$$

Natural Changes in Humidity



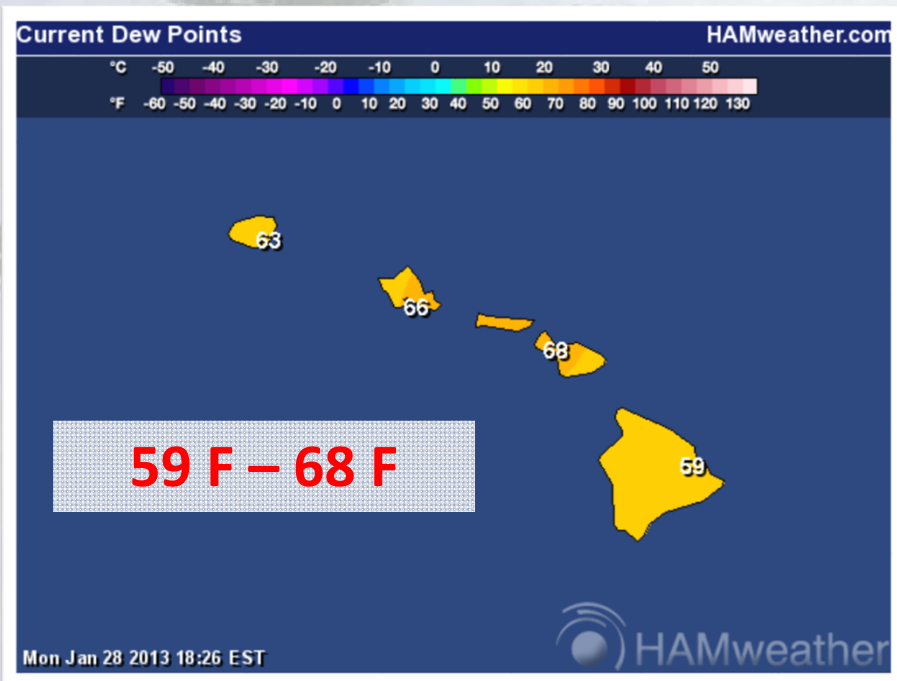
1. Daily changes in temperatures (daylight versus nighttime temperatures)
2. Temperature changes that result as air moves **horizontally** from one location to another
3. Temperature changes caused as air moves **vertically** in the atmosphere





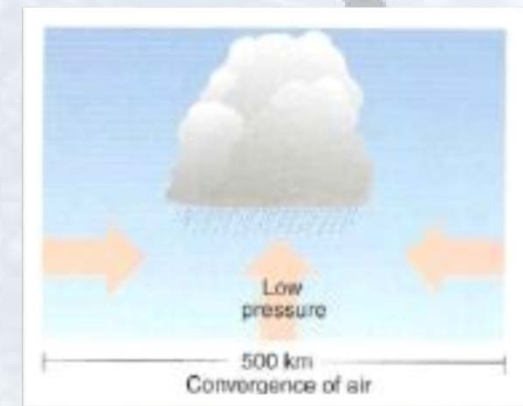
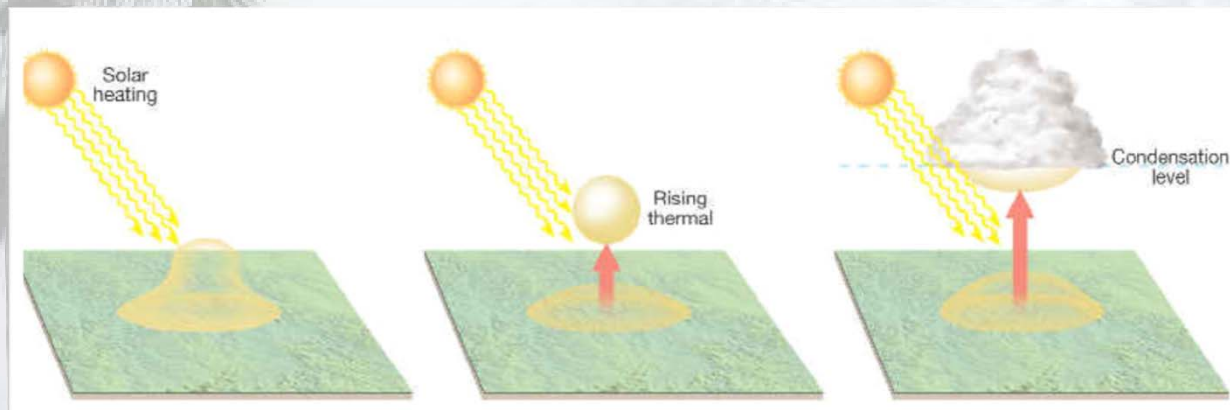
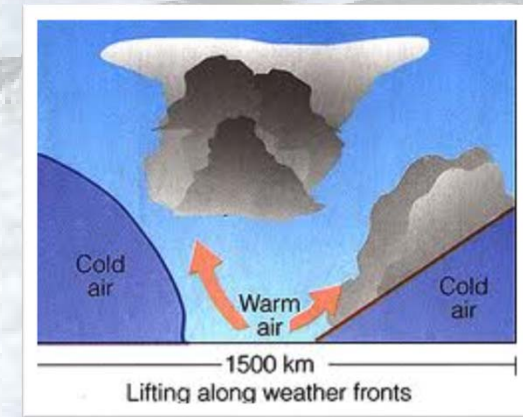
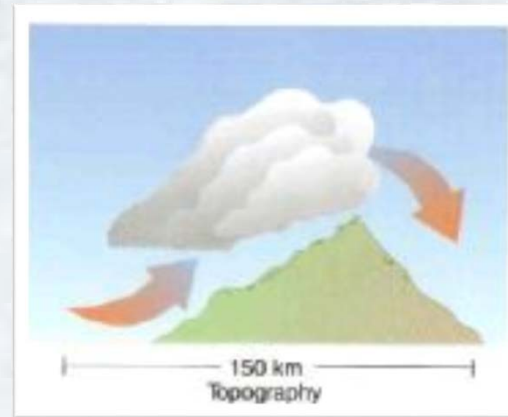
Dew Point Temperature

- The **temperature** at which air needs to be **cooled** to reach **saturation**
- It is a measure of the **actual moisture content** of a parcel of air.
- The term **dew point** stems from the fact that during the night objects at the surface often cool below the dew-point and are coated with dew.
- When the dew point exceeds **~65F** it is considered humid by most people
- A dew point above **75F** is considered unbearable.



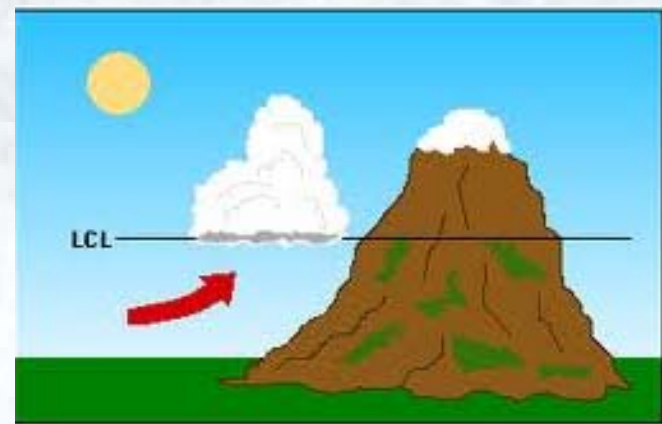
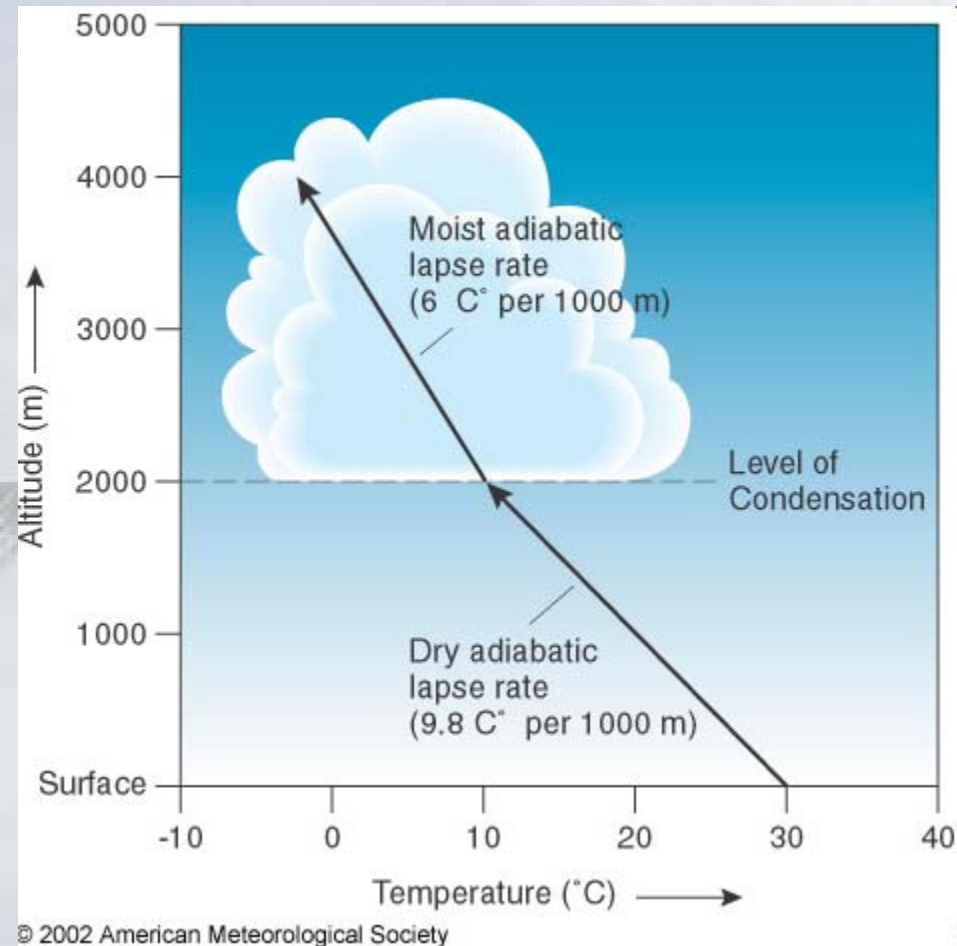
1. Orographic Lifting
2. Frontal Wedging
3. Convergence
4. Localized Convective Lifting (differential heating)

Processes that Lift Air



Lifted Condensation Level (LCL)

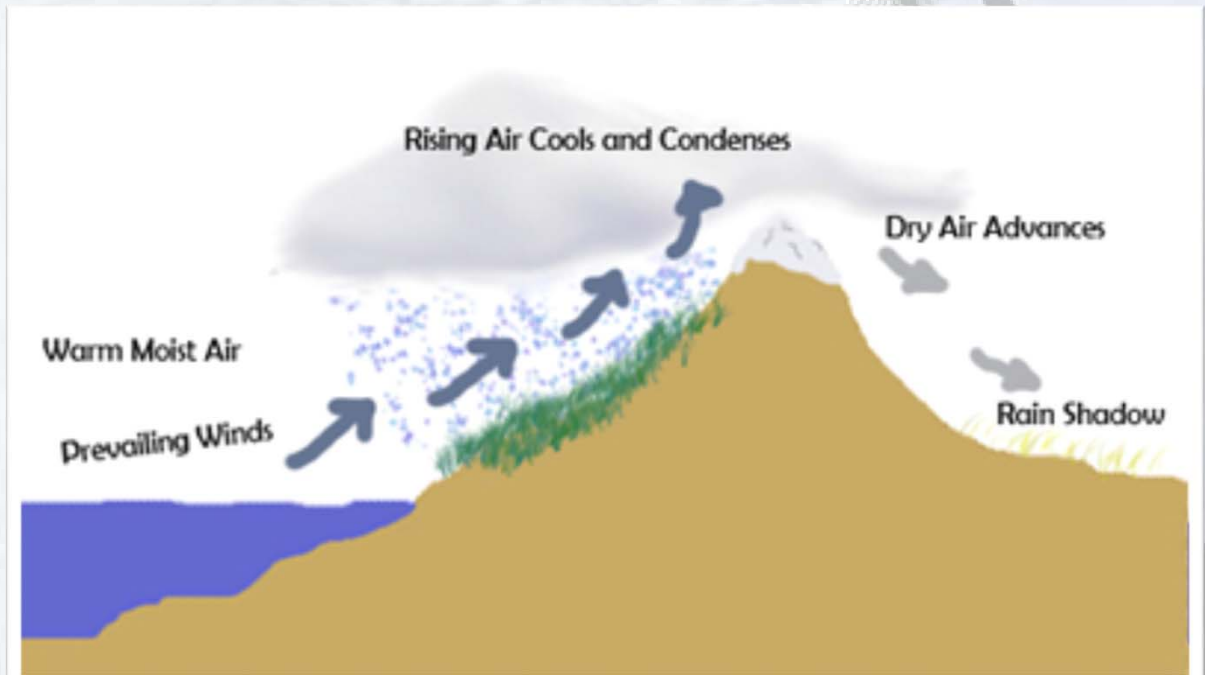
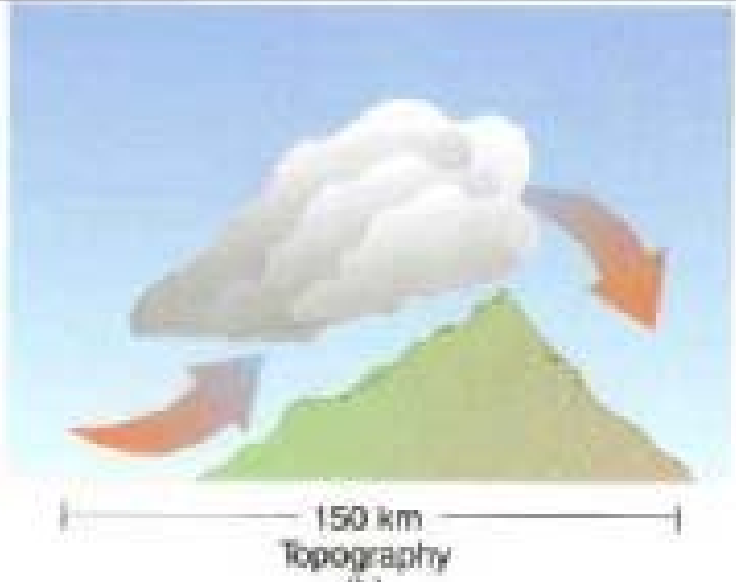
- The height at which rising air that is cooling at the dry adiabatic rate becomes saturated and **condensation** begins.



Why most clouds have FLAT bottoms!!!

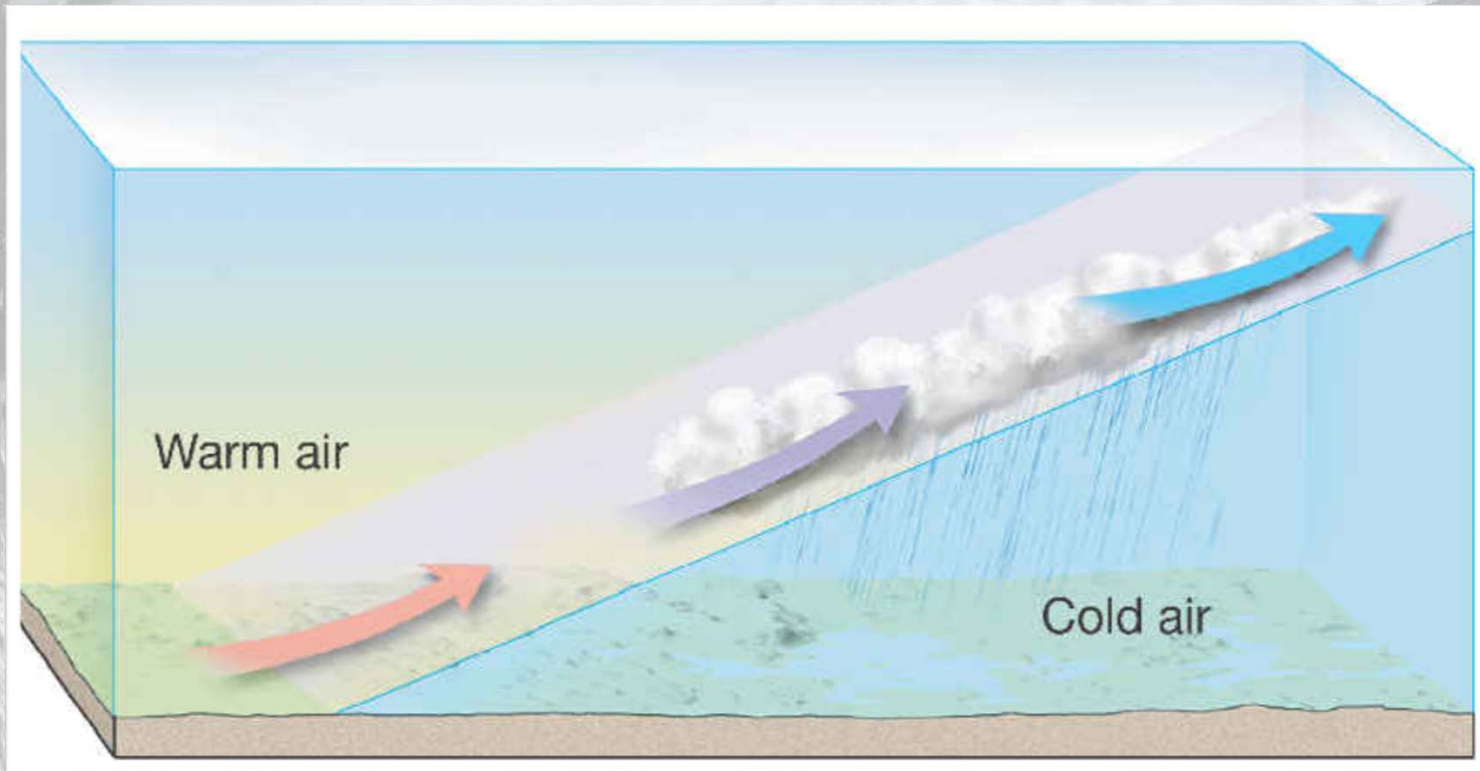
Orographic Lifting

- Air is **forced to rise over** a mountainous or topographic barrier
- Rain shadow desert



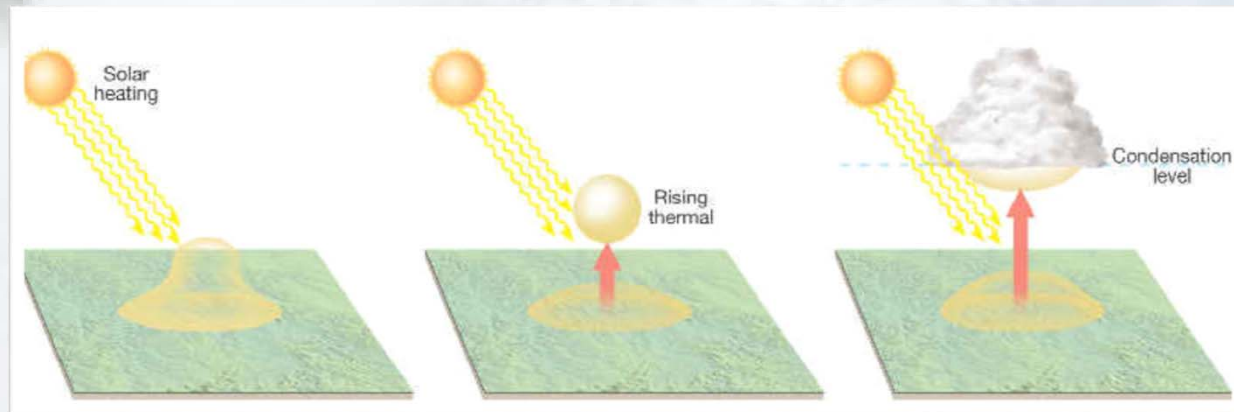
Frontal Wedging

- Warmer, **less dense air**, is forced over cooler, denser air
- **Front** – when warm and cold air collide



Localized Convective Lifting

- **Unequal heating** of Earth's surface causes pockets of air to be warmed more than the surrounding air.
- Buoyant parcels (**thermals**) of hot air rise.
- After reaching the **LCL** they form clouds.



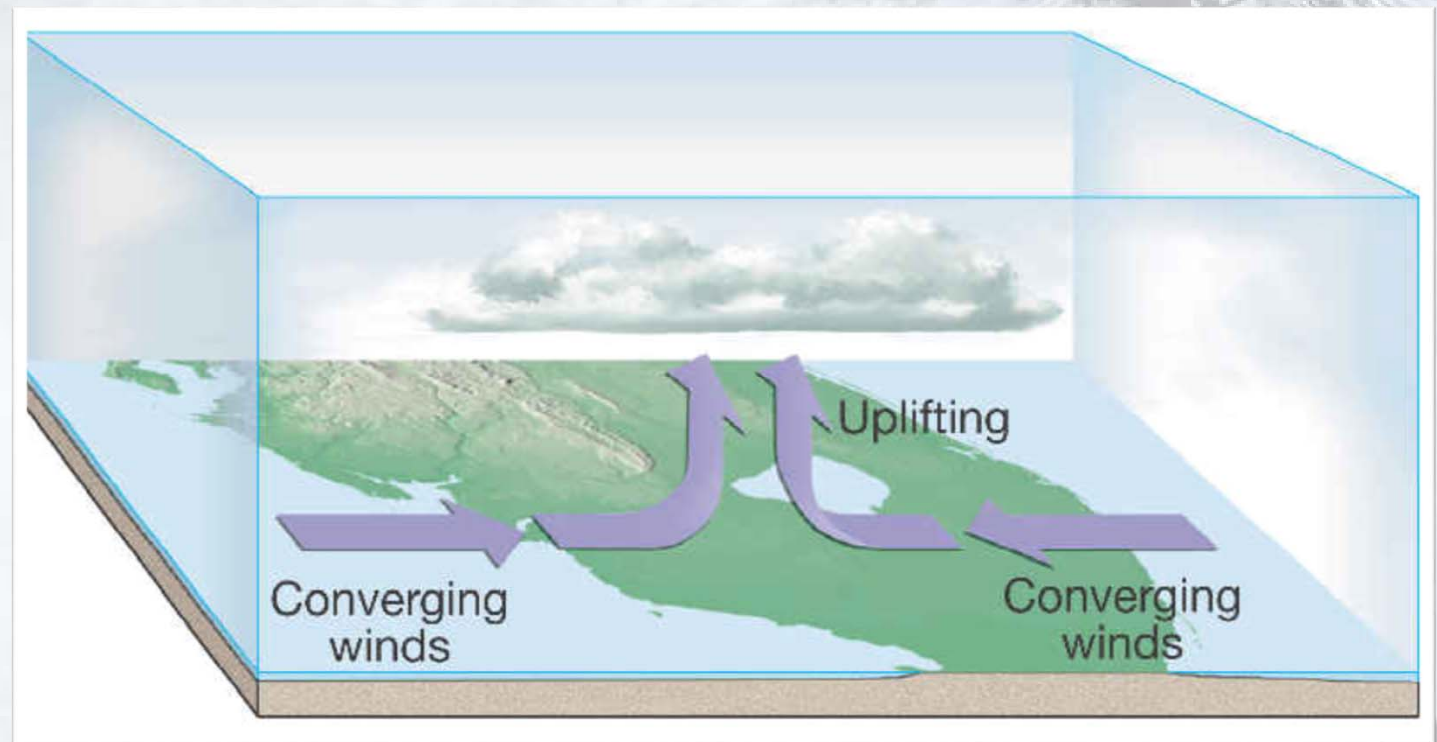
Convergence

When air flows in from **more than one direction** (not a front) can collide

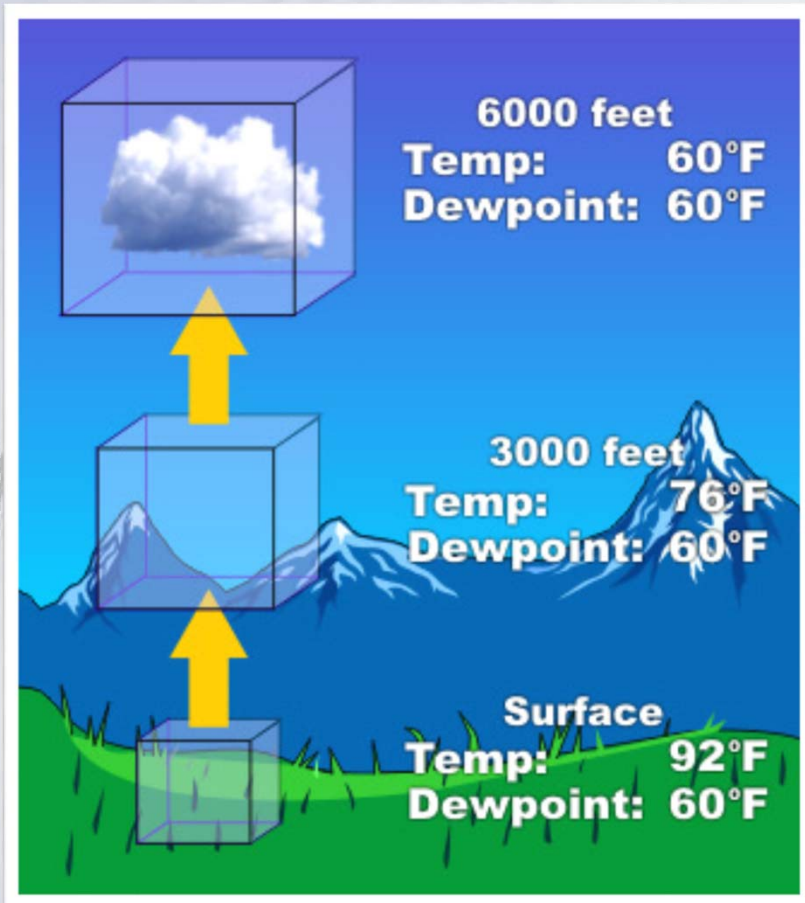
It cannot go down.

It goes up.

Often happens over islands and other regions where two bodies of water are located closely together

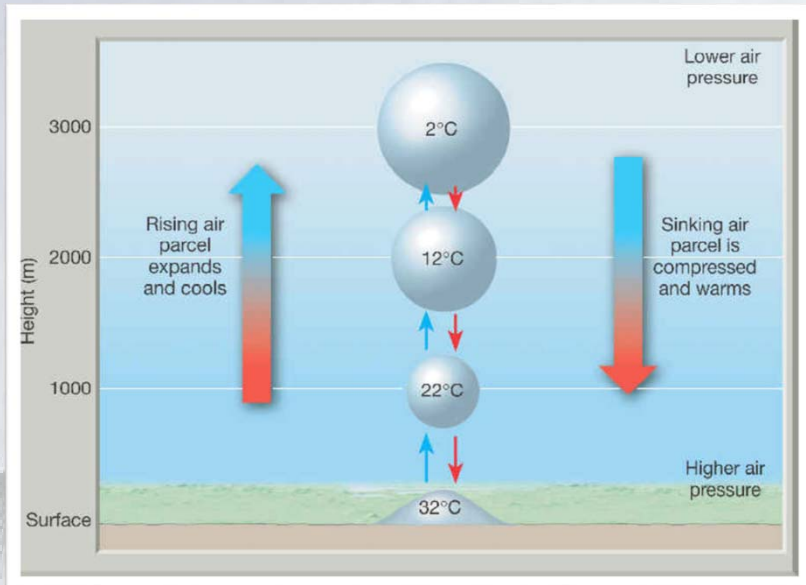


Air Parcels... What are they?



- **A Parcel is an imaginary volume of air**
 - Typically a few hundred cubic meters in volume
 - Acts independently of the surrounding air
 - It is assumed that no heat is transferred into, or out of it
- **HIGHLY IDEALIZED**
- **We use them to talk about the likelihood that air will rise up or sink down.**
 - We need to know this if we want to predict if clouds will form.

Adiabatic Temperature Changes

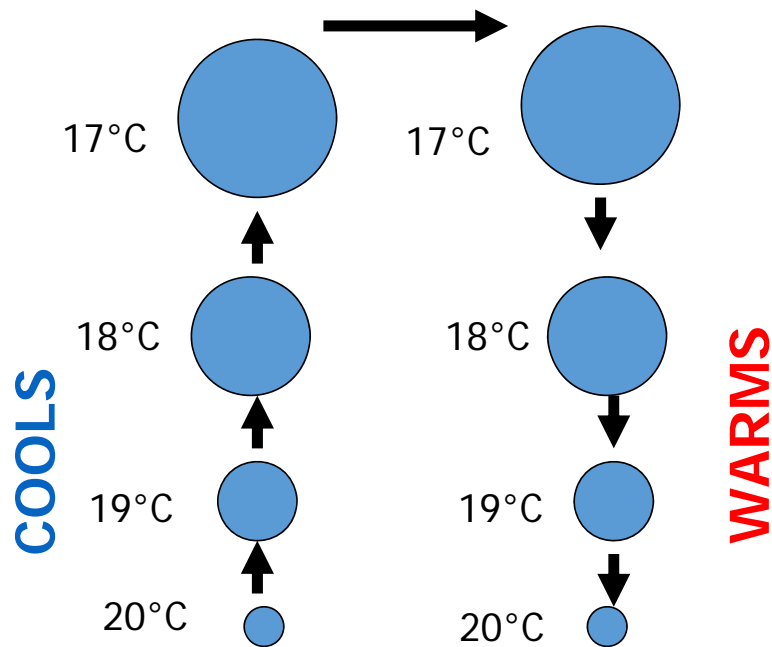


- When heat is neither added nor subtracted
- Result when air is compressed or allowed to expand

When air is allowed to **expand**, it **COOLS**.

When air is **compressed**, it **WARMS**.

Dry Adiabatic Lapse Rate



- The change in temperature due to a change in altitude of a **non-condensing parcel**
- Abbreviated **DALR**
- DALR = $1^{\circ}\text{C}/100\text{m}$
- DALR = $10^{\circ}\text{C}/1000\text{m}$
- DALR = $5.5^{\circ}\text{F}/1000\text{ft}$

Wet Adiabatic Lapse Rate

- The change in temperature due to a change in altitude of a **condensing parcel**

- Abbreviated **WALR**

- WALR = $\sim 0.6^{\circ}\text{C}/100\text{m}$
- WALR = $\sim 6^{\circ}\text{C}/1000\text{m}$
- WALR = $\sim 3.3^{\circ}\text{F}/1000\text{ft}$

